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See also a copy of the **periodic table** with **atomic weights** to five significant figures. See also below the list in Name order. ... 11, Na, **Sodium**, 22.98976928(2). 12, Mg, **Magnesium**, 24.3050(6) ... 19, K, **Potassium**, 39.0983(1), 1. 20, Ca, Calcium, 40.078(4), 1 ... 26, Fe, **Iron**, 55.845(2). 27, Co, Cobalt, 58.933195(5) ... www.chem.gmul.ac.uk/iupac/AfWV/ - Cached - Similar

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Molar mass is the weight of one mole (or 6.02×1023 molecules) of any chemical compounds. ... is found by looking at the **atomic mass** of the element on the **periodic table**. ... (1 atom x 23 grams/mole of **sodium**) + (1 atom x 19 grams/mole of ... + (1 x 1 gram) + (1 x 16 grams) = 56 grams/mole of **potassium** hydroxide ...

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This outcome added to my conviction that the **table** of the **Periodic** ... from a solution of silver nitrate in fused **sodium** and

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Remember changing the number of protons changes the **atomic number**, which is the number of the element. ... Once atomic numbers had significance, the **periodic table** was reorganized. ... the **periodic table**. **Sodium** and **potassium** are examples of these elements. ... Calcium and **magnesium** are examples of alkaline earths. ...

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